SRM VALLIAMMAI ENGINEERING COLLEGE

(An Autonomous Institution)

SRM Nagar, Kattankulathur – 603 203

DEPARTMENT OF PHYSICS

QUESTION BANK



VII SEMESTER

1920703 - ELECTRONICS MATERIALS

Regulation 2019

(ECE, EIE, CSE and Medical Electronics)

Academic Year 2022 – 2023

Prepared by

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DEPARTMENT OF PHYSICS



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SUBJECT: 1920703- ELECTRONICS MATERIALS

SEM / YEAR: VII SEM/AY-2022-2023

UNIT I - INTRODUCTION

Structure: atomic structures and bonding, types of bonding, band formation. Defects and imperfections in solids: Point, Line and Planar defects; Interfacial defects and volume defects. Classification of materials based on bonding: conductors, semiconductors and insulators.

	PART – A				
Q.No	Questions	BT Level	Competence		
1.	Define atomic number and mass number.	BTL 1	Remembering		
2.	Why all atoms are neutral?	BTL 2	Understanding		
3.	Lead has an atomic number of 82 and a mass number of 207. How many protons, neutrons and electrons does it contain?	BTL 3	Applying		
4.	Recall atomic mass unit.	BTL 1	Remembering		
5.	What is an isotope? Give an example.	BTL 2	Understanding		
6.	List the different types of quantum numbers.	BTL 1	Remembering		
7.	State Pauli's exclusion principle.	BTL 1	Remembering		
8.	Recite about valence electrons.	BTL 1	Remembering		
9.	Differentiate electropositive and electronegative elements.	BTL 2	Understanding		
10.	Recall bonding energy.	BTL 1	Remembering		
11.	How is ionic bond formed? Give an example.	BTL 1	Remembering		
12.	Calculate the maximum number of covalent bonds an atom can form.	BTL 3	Applying		
13.	Outline the properties of metallic bond.	BTL 2	Understanding		
14.	What are molecular crystals?	BTL 1	Remembering		
15.	State imperfection.	BTL 1	Remembering		
16.	Report the types of point defects.	BTL 2	Understanding		
17.	Write about Frenkel defect.	BTL 1	Remembering		
18.	Reproduce dislocation.	BTL 1	Remembering		
19.	Define twin boundary.	BTL 2	Understanding		

20.	Recite about energy band.	BTL 1	Remembering
21.	What are forbidden bands?	BTL 1	Remembering
22.	Distinguish conduction band and valence band.	BTL 2	Understanding
23.	Classify materials on the basis of band gap.	BTL 2	Understanding
24.	Give two examples for conductors, semiconductors and insulators.	BTL 2	Understanding
	PART – B		
1.	Define the following.a) Atomb) Atomic Numberc) Mass Numberd) Isotopese) Atomic weightf) Atomic mass unit	BT L 1	Remembering
2.	Analyze the type of bonding that occurs in ionic crystals along with its properties. (13)	BT L 4	Analyzing
3.	Discuss in detail about the bonding that involves the complete transfer of electrons from one atom to another so that each atom acquires a stable electronic configuration similar to the nearest inert gas atoms. (13)	BTL 2	Understanding
4.	With Schematic representation illustrate the formation of bonding involved in CH4 – methane molecule.(13)	BTL4	Analyzing
5.	Illustrate the formation of bonding which involves the mutual sharing of electrons between adjacent atoms each having incomplete outermost shells. (13)	BT L 3	Applying
6.	Interpret why metals are good electrical and thermal conductors. Give your answer in terms of structure and bonding. (13)	BT L 4	Analyzing
7.	Describe the band formation that occurs with the help of delocalized electrons. Also mention the properties of the bond formed. (13)	BT L 2	Understanding
8.	Briefly categorize the main differences between ionic, covalent and metallic bonding. (13)	BTL4	Analyzing
9.	With a neat sketch explain the bonding forces that arises from atomic or molecular dipoles. (13)	BT L 4	Analyzing
10	(i) What is meant by crystal defects? (2) (ii)Explain the various types of one dimensional defects with neat	BT L 1	Remembering
10.	diagram. (11)	BT L 2	Understanding
11.	Discuss in brief about point defects. How these defects affect the properties of materials? (13)	BT L 2	Understanding
12.	With a neat diagram analyze the defects that occurs due to dislocation or distortion of atoms in a line.(13)	BT L 4	Analyzing
13.	Explain in detail the form of different types of dislocations in a crystal and significance of it in determining the properties of the material. (13)	BT L 2	Understanding
14.	Analyze and explain about the various types of interfacial defects occurring in a crystal. (13)	BT L 4	Analyzing
15.	Explain in detail about the various surface defects and its influence in determining the mechanical and physical properties of materials. (13)	BT L 2	Understanding
16.	Outline the formation of energy bands in conductors, semiconductors and insulators. (13)	BT L 4	Analyzing

17.	How do you classify the materials based on the energy gap present in the solid? (13)	BT L 1	Remembering
	PART C		
1.	Elaborate the type of bonding formed with a neat diagram for the following. a) two metals b) two non-metals c) a metal and a non-metal (15)	BTL3	Applying
2.	Report in detail the zero dimensional, one dimensional, two dimensional and three dimensional imperfections in crystals. (15)	BTL3	Applying
3.	Defects are responsible for many of the important properties of materials. interpret the various types of crystal defects and explain their significance in determining the properties for various applications. (15)	BTL4	Analyzing
4.	Isolated atoms have discrete energy levels. When atoms are combined the energy levels overlap and form bands. Outline the formation of bands from discrete energy levels and categorize the materials based on the band formation. (15)	BTL4	Analyzing
5.	Energy levels and bands are the convenient way of representing a solid material. With neat sketch, explain the formation of bands and	BT L 4	Analyzing
5.	classification of solids based on it. (15)		
	classification of solids based on it. (15) UNIT II - CONDUCTING MATERIALS	ased on cor	aductivity of
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13.	What is transition temperature?	BTL 1	Remembering
14.	Infer the effect of magnetic field on the superconducting nature of a material.	BTL 2	Understanding
15.	Superconducting tin has a critical temperature of 3.7 K at zero magnetic field and a critical field of 0.0306 Tesla at 0 K. Calculate the critical field at 2 K.	BTL 3	Applying
16.	Calculate the critical current for a wire of lead having a diameter of 1 mm at 4.2 K. Critical temperature for lead is 7.18 K and $Hc(0) = 6.5 \times 10^4$ A/m.	BTL 3	Applying
17.	The critical temperature for a metal with isotopic mass 199.5 is 4.185 K. calculate the isotopic mass if the critical temperature falls to 4.133 K.	BTL 3	Applying
18.	Plot the magnetization curves for type-I and type-II superconductors.	BTL 3	Applying
19.	Superconductors are perfect diamagnetic materials. Justify.	BTL 2	Understanding
20.	Classify the superconducting materials based on the magnetization behavior in an external magnetic field and based on transition temperature.	BTL 2	Understanding
21.	What are high T_C superconductors? Give any two examples.	BTL 1	Remembering
22.	Write any two medical applications of Superconductors.	BTL 2	Understanding
23.	Give the principle behind magnetic levitated trains.	BTL 2	Understanding
24.	Recall SQUID.	BTL 1	Remembering
	PART – B		
1.	What is electrical conductivity? Mention the factors that affect the conductivity of materials. (13)	BTL 1	Remembering
2.	Categorize the materials based on resistivity. Analyze the various factors that affect the conductivity of materials. (13)	BTL 4	Analyzing
3.	Classify materials based on the conductivity. Explain the temperature dependence of resistivity. (13)	BTL 2	Understanding
4.	 (i) Point out the factors that influence the conductivity of materials. (7) (ii) Illustrate the temperature dependence of resistivity. (6) 	BTL 4	Analyzing
5.	Define low resistive materials. List out the characteristics, properties and applications of any two low resistive materials. (13)	BTL 2	Understanding
6.	Analyze the materials graphite, Al, Cu and steel based on the resistivity. Give the characteristics and properties of each material. (13)	BTL 4	Analyzing
7.	Define High resistive materials. List the characteristics, properties and applications of any two high resistive materials. (13)	BTL 2	Understanding
8.	Infer the resistivity nature of manganin, constantan, nichrome and tungsten. Point out the characteristics and applications of the any two materials. (13)	BTL 4	Analyzing
9.	Explain in detail the properties and applications of any two low and high resistive materials. (13)	BTL 2	Understanding
10.	i) What are superconductors? Give an example.(3)ii) Explain the various properties of superconductor.(10)	BTL 1 BTL 2	Remembering Understanding
11.	Outline the various properties of superconductors. (13)	BTL 4	Analyzing
12.	Write short notes on the following.(4)a) Persistent current(4)b) Meissner effect(5)	BTL 2	Understanding

	c) Effect on current and magnetic field (4)		
13.	i) Explain Meissner effect.(5)ii)Differentiate between type I & type II superconductors.(8)	BTL 2	Understanding
14.	How do you classify the materials based on the response to the applied external magnetic field? Explain them in detail with a neat sketch. (13)	BTL 2	Understanding
15.	(i) Classify the materials based on the critical temperature.(5)(ii) Distinguish between hard and soft superconductors.(8)	BTL 2	Understanding
16.	Explain in detail the applications of superconducting materials. (13)	BTL 2	Understanding
17.	Illustrate the working principles of(6)(i) SQUIDS(6)(ii) Magnetic Levitation(7)	BTL 4	Analyzing
	PART C		
1.	For windings of electrical machines and transformers, copper is preferred to aluminium. Give the characteristics, properties and applications of Cu and Al. List the advantages of Al over Cu in transformer applications. (15)	BTL 4	Analyzing
2.	Enlist some high resistivity alloys suitable as conductive materials. Compare the nichrome, constantan and manganin from different viewpoints. (15)	BTL 4	Analyzing
3.	The resistivity of mercury suddenly drops to zero when cooled below 4.2 K. Identify the phenomenon that arises within it and explain the properties of the above material. (15)	BTL 3	Applying
4.	(i) When a magnetic field is applied, diamagnetic materials repel back all the lines of force. Superconductors are perfect diamagnetic material. Explain the diamagnetic nature of superconductors. Explain the application which works on this diamagnetic behavior of superconductors. (10) (ii) Prove that the susceptibility of a superconductor is -1 . (5)	BTL 3	Applying
5.	Discuss the applications of superconducting materials in the following (i) Fast electrical switching (ii) Detection of very weak magnetic signals (iii) Super-fast trains without frictional loss (15)	BTL 3	Applying

UNIT III - SEMICONDUCTING AND MAGNETIC MATERIALS

Semiconductors: Introduction, types of semiconductors, temperature dependence of semiconductors, compound semiconductors, basic ideas of amorphous and organic semiconductors. Magnetic Materials: classification of magnetic materials, ferromagnetism-B-H curve (Qualitative), hard and soft magnetic materials, magneto materials applications.

	$\mathbf{PART} - \mathbf{A}$		
Q.No	Questions	BT Level	Competence
1.	Define semiconductor in terms of band gap.	BTL 1	Remembering
2.	List the properties of semiconductors.	BTL 2	Understanding

3.	What are elemental and compound semiconductors?	BTL 1	Remembering
4.	Classify the semiconductors based on the impurities present in the material. Give two examples of each semiconductors.	BTL 2	Understanding
5.	Give the limitations of the intrinsic semiconductors.	BTL 2	Understanding
6.	Recall doping.	BTL 1	Remembering
7.	How is an n-type semiconductor formed and give an example?	BTL 2	Understanding
8.	Sketch the energy band diagram of an extrinsic semiconductor.	BTL3	Applying
9.	Differentiate intrinsic and extrinsic semiconductors.	BTL 2	Understanding
10.	Recite about organic semiconductors.	BTL 1	Remembering
11.	Define amorphous semiconductors.	BTL 1	Remembering
12.	Define magnetic dipole.	BTL 1	Remembering
13.	What is intensity of magnetization?	BTL 1	Remembering
14.	Establish the relation between relative permeability (μ_r) and susceptibility (χ) .	BTL3	Applying
15.	Categorize the magnetic materials based on magnetic moment.	BTL 2	Understanding
16.	Why diamagnets are called weak magnets?	BTL 1	Remembering
17.	Magnetic field Intensity of a paramagnetic material is 10^4 A/m. At room temperature, its susceptibility is 3.7×10^{-3} . Calculate the magnetization in the material.	BTL 3	Applying
18.	The magnetic susceptibility of a medium is 940 x 10^{-4} . Calculate its relative permeability.	BTL 3	Applying
19.	Recall magnetic domains.	BTL 1	Remembering
20.	Mention the four types of energies involved in the growth of magnetic domains.	BTL 2	Understanding
21.	Recall the term retentivity.	BTL 2	Understanding
22.	Interpret hysteresis loss.	BTL 1	Remembering
23.	Differentiate soft and hard magnetic materials.	BTL 2	Understanding
24.	Define Giant Magneto-resistance.	BTL 1	Remembering
	PART – B		
1.	What are semiconductors? Classify the semiconductors based on composition of the material, impurities and band gap. (13)	BTL 1	Remembering
2.	(i) Compare and contrast between elemental and compound semiconductors.(8)(ii) Infer the temperature dependence of semiconductors.(5)	BTL 4	Analyzing
3.	What is an extrinsic semiconductor? Sketch the energy band diagram and explain n-type and p-type semiconductor with an example. $(3 + 10)$	BTL 2	Understanding
4.	(i) Distinguish between intrinsic semiconductors and extrinsic semiconductors.(6)(ii) Compare and contrast n-type and p-type semiconductor.(7)	BTL 4	Analyzing

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5.	(i) Discuss about compound semiconductors?(6)(ii) Distinguish between elemental and compound semiconductors.(7)	BTL 2	Understanding
6.	Write short notes on the following(4)(i) Compound semiconductors(4)(ii) Organic Semiconductors(4)(iii) Amorphous Semiconductors(5)	BTL 1	Remembering
7.	Give the basic ideas about organic and amorphous semiconductors. (13)	BTL 1	Remembering
8.	Explain the classification of magnetic materials based on their spins. (13)	BTL 2	Understanding
9.	Compare and contrast the characteristics and properties of diamagnetic, paramagnetic and ferromagnetic material. (13)	BTL 4	Analyzing
10.	 (i)What is meant by Ferromagnetism? (3) (ii) Explain with neat diagrams, how the magnetization occurs in a ferromagnetic magnetic material, after the application of magnetic field. 	BTL 2	Understanding
11.	Examine the different types of energy involved in domain growth. (13)	BTL 4	Analyzing
12.	Explain the different types of energies that are responsible for growth of domains in a ferromagnetic material. (13)	BTL 4	Analyzing
13.	 (i) Define the term coercivity and retentivity. (3) (ii)Explain the lagging of magnetization behind the magnetizing field with the help of an B-H curve. (10) 	BTL 2	Understanding
14.	(i)What are soft and hard magnetic materials? (3)(ii)Write the differences between soft and hard magnetic materials. (10)	BTL 1	Remembering
15.	(i)Write a short note on B-H curve.(6)(ii)Differentiate soft and hard magnetic materials.(7)	BTL 2	Understanding
16.	Demonstrate the working of magnetic hard disc based on GMR sensor. Mention its advantages and disadvantages. (13)	BTL 3	Applying
17.	Examine the writing and reading of data in magnetic hard disk using GMR sensors. (13)	BTL 4	Analyzing
	PART C		
1.	Si is used in manufacturing of integrated chips and different electronic components. With a neat diagram, explain about the formation of pure silicon and the formation of n-type and p-type with the help of impurities. (15)	BTL 3	Applying
2.	Spin of the magnetic materials and its arrangement decides the properties of magnetic materials. With necessary diagrams and examples, compare the properties of magnetic materials based on their spins. (15)	BTL 3	Applying
3.	Ferromagnetic magnetic material is similar to a polycrystalline material with a large number of domains in it. Discuss the spontaneous magnetization of ferromagnetic materials with the help of domains. Compile the theory based on energy involved in the ferromagnetic material. (15)	BTL 4	Analyzing
4.	When a ferromagnetic material is placed in a magnetic field (H), the magnetic induction (B) lags behind applied field. Plot an B – H curve and explain the magnetic hysteresis loop formation on the basis of domain theory of ferromagnetism. (15)	BTL 3	Applying

5. GMR sensors which have a very high magnetic sensitivity, are used to read the data at greater speed. Explain the process of recording binary data (0 and 1) and the process of retrieving the information with the help of a GMR sensor. (15)

UNIT IV - DIELECTRIC AND INSULATING MATERIALS

Dielectric Materials: Introduction, classification, temperature dependence on polarization, properties, dielectric loss, factors influencing dielectric strength and capacitor materials, applications. Insulators: Introduction, thermal and mechanical properties required for insulators, Inorganic materials, organic materials, liquid insulators, gaseous insulators and ageing of insulators, applications.

Q. No	Questions	BT Level	Competence
1.	Define dipole moment.	BTL 1	Remembering
2.	Recall dielectric constant.	BTL 1	Remembering
3.	What is meant by polarization in a dielectric material?	BTL 1	Remembering
4.	Categorize the different types of polarization.	BTL 2	Understanding
5.	Recite ionic polarization.	BTL 1	Remembering
6.	What is space charge polarization?	BTL 1	Remembering
7.	Calculate the electronic polarizability of an isolated Se atom. The atomic radius of an Se atom is 0.12 nm.	BTL 3	Applying
8.	What is the polarization produced in sodium chloride by an electric field of 600 V/mm if it has a dielectric constant of 6?	BTL 3	Applying
9.	If NaCl is subjected to an electrical field of 1000 V m^{-1} and the resulting polarization is 4.3 X 10^{-8} C/m^2 , calculate the relative permittivity of NaCl.	BTL 3	Applying
10.	Interpret the effect the temperature on orientational and ionic polarization.	BTL 2	Understanding
11.	Classify the different types of dielectric materials.	BTL 2	Understanding
12.	Indicate the factors affecting dielectric loss.	BTL 2	Understanding
13.	Define dielectric strength.	BTL 1	Remembering
14.	Identify the properties of dielectrics in capacitor applications.	BTL 2	Understanding
15.	Mention any four uses of dielectrics in transformers.	BTL 1	Remembering
16.	How does the porosity affects the mechanical properties of an insulating material?	BTL 2	Understanding
17.	Outline the effect of thermal conductivity on insulating materials.	BTL 2	Understanding
18.	Categorize the classification of insulators based on the material used.	BTL 2	Understanding
19.	Name some of the organic and inorganic insulating materials.	BTL 1	Remembering
20.	What are askarels?	BTL 2	Understanding
21.	Give the characteristics of silicone fluids.	BTL 2	Understanding
22.	Summarize the advantages of inorganic fiber materials over organic	BTL 2	Understanding

PART	$-\mathbf{A}$
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	fibers materials.		
23.	Recall ageing in insulators.	BTL 1	Remembering
24.	Write any four applications of insulators.	BTL 1	Remembering
	PART – B		
1.	Define polarization. Discuss the different types of polarization and the effect of temperature on these polarizations. (3+10)	BTL 2	Understanding
2.	Examine the various types of polarization mechanism involved in dielectrics and infer their dependence of temperature. (13)	BTL 3	Applying
3.	What is dielectric loss? Derive an expression for power loss and the factors affecting dielectric loss.(3 +10)	BTL 2	Understanding
4.	Outline the phenomenon of dielectric loss and derive an expression for it. Give the various factors affecting the power loss in dielectrics. (13)	BTL 2	Understanding
5.	 (i) Define dielectric strength and analyze the factors that influence the dielectric strength of a material. (8) (ii)Predict the effect of temperature on different polarization mechanism. (5) 	BTL 4	Analyzing
6.	Discuss the uses of dielectric materials as capacitors and transformers. (13)	BTL 2	Understanding
7.	Explain in detail about the application of dielectrics. (13)	BTL 1	Remembering
8.	Define insulators. Interpret the thermal and mechanical properties of an insulating material. (3+10)	BTL 3	Applying
9.	Describe in detail about the various solid insulating materials. (13)	BTL 2	Understanding
10.	Discuss in detail about the organic and inorganic materials used as solid insulating materials. (13)	BTL 2	Understanding
11.	Describe in detail about the various liquid insulation materials. (13)	BTL 2	Understanding
12.	Explain in detail the properties and characteristics of various liquids as an insulating material. (13)	BTL 1	Remembering
13.	List and explain the properties and applications of gases in insulating materials. (13)	BTL 1	Remembering
14.	Write briefly about the gaseous insulating materials. (13)	BTL 2	Understanding
15.	 (i) Analyze the thermal and mechanical properties of an insulating material. (ii) Determine the various ageing phenomenon that can take place in insulators. (5) 	BTL 4	Analyzing
16.	What is ageing? Explain about the various ageing process that can occur in insulation materials. List the remedies to avoid ageing in insulators. (13)	BTL 1	Remembering
17.	(i) Outline the phenomenon of ageing in insulators.(6)(ii) Discuss the applications of insulators.(7)10	BTL 2	Understanding

	PART - C		
1.	The displacement between positively and negatively charged particles by a small distance on the application of electric field produces electric dipoles. Discuss the polarization phenomenon occurring in various dielectric and the effect of temperature on these polarizations. (15)	BTL 4	Analyzing
2.	The power loss in an ideal capacitor is zero, but it is difficult to manufacture ideal capacitors. Explain the power loss in dielectric materials when an ac voltage is applied and arrive an expression for the power loss in commercial dielectrics. (15)	BTL 4	Analyzing
3.	Identify the properties of dielectric that makes it suitable for capacitor and transformer applications. Discuss the applications of dielectrics in capacitors and transformers. (15)	BTL 3	Applying
4.	Describe in detail about the different types of insulating materials along with their characteristics and applications. (15)	BTL 2	Understanding
5.	(i) Explain the thermal and mechanical properties of insulating materials. (10) (ii) Mention the applications of insulating materials. (5)	BTL 1	Remembering

UNIT V - OPTOELECTRONIC AND NANO ELECTRONIC MATERIALS

Optoelectronic materials. Introduction, properties, factor affecting optical properties, role of optoelectronic materials in LEDs, LASERs, photo detectors, solar cells. Nano electronic Materials: Introduction, advantage of nanoelectronic devices, materials, fabrication, challenges in Nano electronic materials.

Q.No	Questions	BT Level	Competence
1.	What are optical materials?	BTL 1	Remembering
2.	Classify the optical materials based on their interaction with visible light.	BTL 2	Understanding
3.	Recall translucent materials.	BTL 1	Remembering
4.	Interpret the opaque nature of metals.	BTL 3	Applying
5.	Are all insulators transparent in nature? Justify.	BTL 3	Applying
6.	What are optoelectronic devices?	BTL 1	Remembering
7.	Justify why LEDs are preferred to have hemispherical shape?	BTL 3	Applying
8.	Calculate the wavelength of radiation emitted by an LED with band gap energy 2.8 eV.	BTL 3	Applying
9.	Identify the suitable material for producing blue LEDs? Explain why?	BTL 2	Understanding
10.	What is a heterojunction laser?	BTL 1	Remembering
11.	Summarize the applications of laser diodes.	BTL 2	Understanding
12.	Define photodetectors.	BTL 1	Remembering
13.	Write the working principle of solar cell.	BTL 2	Understanding
14.	Mention any two merits of solar cell.	BTL 1	Remembering
15.	The wavelength of light emission in an LED is 1.55 µm. Calculate the band gap in eV?	BTL 3	Applying

PART - A

16.	State Moore' law.	BTL 1	Remembering
17.	Identify the major drawbacks of microelectronics.	BTL 2	Understanding
18.	Define nanoelectronics.	BTL 1	Remembering
19.	Give any four advantages of nanoelectronics devices.	BTL 1	Remembering
20.	List some of the commonly used materials in nanoelectronics devices.	BTL 2	Understanding
21.	What is top-down and bottom-up approach of nanomaterials synthesis?	BTL 1	Remembering
22.	Mention any two advantages of ball milling method.	BTL 2	Understanding
23.	Recite about lithography.	BTL 1	Remembering
24.	Write any two applications of sol-gel method.	BTL 1	Remembering
	PART – B		
1.	Enlist various optical phenomena that are observed when light falls on a material? Briefly explain each of them. Explain: reflectivity, absorptivity and transmissivity. (13)	BTL 2	Understanding
2.	Differentiate between transparent, translucent and opaque materials. Quote examples of each of them. (13)	BTL 2	Understanding
3.	Outline the optical properties of non-metals, metals, semiconductors and insulators. Why does the gold appear yellow, but silver appears white? (13)	BTL 4	Analyzing
4.	What is meant by minority charge carrier injection? Explain how a P-N junction diode acts as a LED.(13)	BTL 2	Understanding
5.	Describe the principle and working of LED with a neat diagram and mention its advantages and disadvantages. (13)	BTL 2	Understanding
6.	What is a laser diode? Explain the construction and working of Ga-As laser. (13)	BTL 1	Remembering
7.	Discuss the principle, construction and working of a semiconductor diode laser. (13)	BTL 2	Understanding
8.	Demonstrate the principle, construction and working of a solar cell with a neat diagram. Also mention any four applications. (13)	BTL 3	Applying
9.	Explain the principle and working of a large area photo-voltaic device which converts sunlight directly into electricity. (13)	BTL 2	Understanding
10.	Illustrate the construction and working of a p-i-n photo detector. (13)	BTL 4	Analyzing
11.	Analyze the construction and working of an avalanche photo detector. (13)	BTL 4	Analyzing
12.	 (i) What is nanoelectronics? State Moore's law and the need for nanotechnology in electronics. (5) (ii) List the advantages of nanoelectronics devices. 	BTL 1	Remembering
13.	With a neat sketch demonstrate how the nanoparticles are prepared employing transfer of design of patterns on a semiconductor substrate by lithography. (13)	BTL 3	Applying
14.	Describe the conventional method of particle size reduction by mechanical crushing. (13)	BTL 2	Understanding
15.	Explain in detail about chemical vapor deposition technique for fabrication of nanomaterials. (13)	BTL 2	Understanding

16.	Explain the synthesis of nanoparticles using sol-gel method. (13)	BTL 2	Understanding			
17.	Describe any one method for the synthesis of nanomaterials. (13)	BTL 2	Understanding			
PART-C						
1.	Explain the principle of injection luminescence. Describe the theory and working of LED. What are the advantages of using LED in electronic display? (15)	BTL 2	Understanding			
2.	Can light be used in materials processing like cutting, welding etc. List the properties to be modified in a light to make it suitable for material processing. With necessary description and explain the construction and working of a solid state laser source used for cutting complex shapes. (15)	BTL 3	Applying			
3.	Communication satellites require an electric power source to work in the vacuum of space. Design a photovoltaic cell circuit using a semiconducting material to power the satellites. With a neat diagram explain the operation of the above circuit when exposed to sunlight along with its advantages and disadvantages. (15)	BTL 3	Applying			
4.	A detector circuit in the street lights uses switch –on relays to turn on the street lights and turn off the lights when the diode fails to conduct. Construct a circuit for the above said detector and explain its operation along with the advantages and disadvantages. (15)	BTL 3	Applying			
5.	Describe a technique to synthesize nanophase materials. Discuss their applications in various fields. (15)	BTL 2	Understanding			